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# The structure of $\beta$ -U<sub>3</sub>O<sub>8</sub>. By B.O. LOOPSTRA, Reactor Centrum Nederland, Petten, The Netherlands

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The structure of  $\beta$ -U<sub>3</sub>O<sub>8</sub> has been determined from neutron-diffraction powder data. The oxygen coordination is a pentagonal dipyramid for two uranium atoms; for the third it is a deformed octahedron.

At room temperature two crystalline forms of  $U_3O_8$  are known. Normally the  $\alpha$ -form is encountered, but under special circumstances the  $\beta$ -form is obtained (Hoekstra, Siegel, Fuchs & Katz, 1955). Unit-cell data for the two substances suggest a close structural relationship (Table 1). In this paper a structure determination of  $\beta$ -U<sub>3</sub>O<sub>8</sub> is presented to permit a comparison with the structure of  $\alpha$ -U<sub>3</sub>O<sub>8</sub> (Loopstra, 1962).

 $\beta$ -U<sub>3</sub>O<sub>8</sub> was prepared by heating  $\alpha$ -U<sub>3</sub>O<sub>8</sub> to 1350 °C in air or oxygen, followed by slow cooling (100 °C per day) to room temperature (Cordfunke, 1969). A coarsely crystal-line product was obtained.

Neutron-diffraction powder data were collected at room temperature from a sample, contained in a thin-walled vanadium sample holder of 15 mm diameter. A wavelength of 2.5776(5) Å was employed. The range of observation was  $0 < (\sin \theta)/\lambda < 0.361 \text{ Å}^{-1}$ , containing 85 possible reflexions. Soller slits of 10' divergence were placed between Table 1. Unit-cell data of  $\alpha$ -U<sub>3</sub>O<sub>8</sub> and  $\beta$ -U<sub>3</sub>O<sub>8</sub>, obtained from X-ray powder diagrams

Primed axes refer to the subcell. The value of a'/b' indicates the deviation from hexagonality (ideal value: 1.7321).

Phase	α	β
Space group	$C2mm(C_{2v}^{14})$	$Cmcm(D_{2b}^{17})$
a	6·716 (1) Å	7·069 (1) Å
b	11.960 (2)	11.445 (1)
с	4.1469 (5)	8.303 (1)
Z	2 .	4
a'	6·716 (1) Å	7·069 (1) Å
b'	3.9868 (6)	3.8151 (5)
c'	4.1469 (5)	4.1515 (5)
a'/b'	1.6846(2)	1.8529 (2)

the H.F.R. reactor and the Cu(111) monochromator, and between the sample and the detector.



Fig. 1. Observed and calculated powder pattern. Dots: observed data, corrected for background. Full curve: calculated pattern.



Fig. 2. Relationship between sections perpendicular to the c axis of  $\alpha$ -U<sub>3</sub>O<sub>8</sub> (a), 'ideal' UO<sub>3</sub> (b), and  $\beta$ -U<sub>3</sub>O<sub>8</sub> (c). In (a) and (c) dots represent the actual positions, isolated dots uranium atoms, and dots connected by lines oxygen atoms; circles represent 'ideal' positions, small circles uranium atoms, and large circles oxygen atoms. Ideal positions are obtained from (b) by removing the right-hand atom from each pair of double-circled oxygen atoms to obtain  $\alpha$ -U<sub>3</sub>O<sub>8</sub>, while for  $\beta$ -U<sub>3</sub>O<sub>8</sub> each pair is replaced by one atom halfway between. Note shift of origin by b/3 in (c). In (a) the section is at z=0, in (c) it is at  $z=\frac{1}{2}$ .

It was found that the neutron diffraction pattern was clearly related to that of  $\alpha$ -U<sub>3</sub>O<sub>8</sub>. Therefore a trial structure was sought which closely resembled the structure of  $\alpha$ -U<sub>3</sub>O<sub>8</sub> (Loopstra, 1962), but which at the same time would explain the occurrence of several sizable reflexions with  $l \neq 2n$ . Some models that readily suggested themselves were refined by preliminary least-squares computations, based on integral peak intensities. One model, which turned out to give good agreement, was refined by use of the profile-fitting technique described by Rietveld (1969). The resulting fit is displayed in Fig.1. The final values of the parameters and their standard deviations are given in Table 2.

#### Table 2. Final least-squares parameters

		x	У	Z
U(1)	4( <i>c</i> )	0 (-)	0.989 (1)	0.25 (-)
U(2)	4(c)	0 (-)	0.350(1)	0.25 (-)
U(3)	4(c)	0 (-)	0.668(1)	0.25 (-)
O(1)	4(a)	0 (-)	0 (-)	0 (-)
O(2)	8(f)	0 (-)	0.352 (1)	0.023 (1)
O(3)	4(c)	0 (-)	0.165 (1)	0.25 (-)
O(4)	8(g)	0.318 (1)	0.024 (1)	0.25 ()
O(5)	8(g)	0.319(1)	0.312 (1)	0.25 (-)

In Fig. 2 the structural relationship between  $\alpha$ - and  $\beta$ -U<sub>3</sub>O<sub>8</sub> is visualized. From a layer of hypothetical 'ideal' UO<sub>3</sub> [Fig. 2(*b*)] the idealized  $\alpha$  form is derived by removing one oxygen atom from every third row as indicated in Fig. 2(*a*). (Holmberg, 1970). To obtain the idealized  $\beta$  form two oxygen atoms are replaced by one halfway between them [Fig. 2(*c*)]. Fig.2(*a*) and (*c*) shows that the actual structures are only slightly distorted from the hypothetical ideal structures. In  $\beta$ -U<sub>3</sub>O<sub>8</sub> the layers are stacked along the *c* axis such that a set of chains of U atoms, surrounded by oxygen atoms at the corners of a pentagonal dipyramid, is formed at x=0, y=0, and  $x=\frac{1}{2}$ ,  $y=\frac{1}{2}$ . The other uranium atoms form chains in the *c* direction in which the oxygen coordination is alternately pentagonal dipyramidal and distorted octahedral. All nearest neighbour groups of an octahedrally coordinated U atom are pentagonal dipyramidal. This situation may be compared with that in  $\alpha$ -U<sub>3</sub>O<sub>8</sub>, where all uranium atoms are pentagonal dipyramidally coordinated.

In Table 3 the U–O distances shorter than 3.5 Å have been collected. From these it may be seen that the oxygen bonding around U(3) cannot be described in terms of a uranyl group, in which case two distances would be considerably shorter than the others. It has been checked that this result is not due to a wrong choice of the z parameter of O(2): refinement starting from z = -0.023 leads to a positive parameter in a few cycles.

## Table 3. Coordination of U atoms

	Number of equivalent	
Atom(1)-atom(2)	distances	Distance
U(1)-O(1)	2	2·080 (1) Å
U(1) - O(3)	1	2.021(21)
U(1) - O(4)	2	2.287 (10)
U(1) - O(5)	2	2.398 (14)
U(2) - O(2)	2	1.888 (8)
U(2) - O(3)	1	2.112 (21)
U(2) - O(4)	2	2.368 (13)
U(2) - O(5)	2	2.295 (10)
U(3) - O(2)	2	2.275 (8)
U(3) - O(4)	2	2.087 (16)
U(3) - O(5)	2	2.088 (16)

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